

## STRUCTURAL STUDY OF TUBEIMOSIDE I, A CONSTITUENT OF TU-BEI-MU

Fan-hua Kong, Da-yuan Zhu, Ren-sheng Xu\*  
Shanghai Institute of Materia Medica, Academia Sinica,  
319 Yueyang Road, Shanghai 200031, CHINA  
Zhang-cai Fu, Lang-yin Zhou  
Second Hospital of Xian Medical College,  
36 Middle Xiwu Road, Xian, CHINA  
Takashi Iwashita, Hajime Komura  
Suntary Institute for Bioorganic Research(SUNBOR),  
Wakayamadai, Mishima-gun, Osaka 618, JAPAN

The structure of Tubeimoside I isolated from the bulb of *Bolbostemma paniculatum* [Maxim] Franquet (Tu-bei-mu) was determined as **1** mainly by comparative NMR studies of **1** and its degradation products. The compound **1** has inter-saccharide chain bridging by a dicrotalic acid to form a unique macrocyclic structure.

Tu-bei-mu, the bulb of *Bolbostemma paniculatum* [Maxim] Franquet, is one of the Chinese folk medicines often used for the treatment of tumors as well as for detoxication. It has also been used to cure warts.<sup>1,2</sup> Maltose has been isolated from these bulbs, but no other chemical components have been isolated and identified.<sup>3</sup> We have isolated and identified seven chemical constituents of Tu-bei-mu including maltol and  $\Delta^{7,16,25(26)}$ -stigmastatrien-3-ol. The structure of one of the new triterpenoid saponins, tubeimoside I<sup>4</sup> **1**, which has an unusual dicrotalic acid bridging between two sugar chains, was determined as follows.

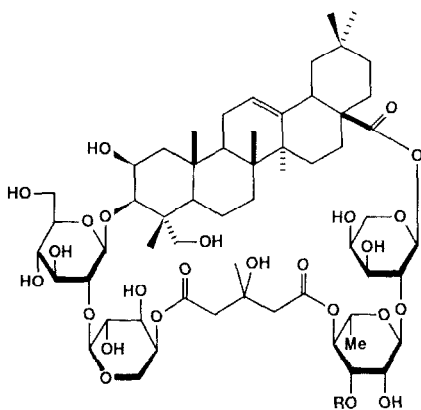
Tubeimoside I, white needles, mp; 250-2°C,  $[\alpha]_D +13.0^\circ$  (c 0.8, MeOH), has the molecular formula  $C_{63}H_{98}O_{29}$ , MW 1318 (SIMS;  $m/z$  1319, M+H, and 1341, M+Na). Mild acid hydrolysis of **1** gave bayogenin<sup>5</sup> and 4 sugars: L-arabinose, D-glucose, L-rhamnose and D-xylose. However, the  $^{13}C$ -NMR spectrum of **1** (Table 1), shows five sugar anomeric carbons. The presence of a dicrotalic acid moiety<sup>6</sup> was also suggested from the NMR data ( $^1H$ -NMR:  $\delta$ 2.03, 3H, s; 3.46 and 3.05, 1H each, AB-system,  $J=15.8$ Hz; 3.31 and 3.07, 1H each, AB-system,  $J=15.6$ Hz;  $^{13}C$ -NMR: see Table 1). The signals at  $\delta$ 94.1 and 175.9 indicated the presence of a glycosidic ester linkage at C-28 of bayogenin.<sup>7</sup>

Hydrolysis of **1** with 1N KOH afforded prosapogenin **2** (FDMS,  $m/z$  783, M+H) and trisaccharide **3**, which was originally linked to C-28. The prosapogenin **2** was subjected to acid hydrolysis to give bayogenin, arabinose and glucose, while the trisaccharide **3** gave arabinose, rhamnose and xylose. By comparison of the  $^{13}C$ -NMR spectrum of **2** with those of bayogenin methyl ester and methyl  $\beta$ -D-glucoside and

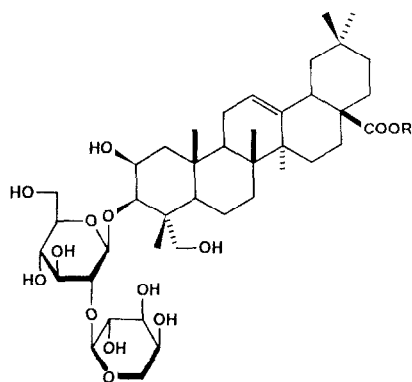
$\alpha$ -L-arabinopyranosides,<sup>8</sup> two glycosidation shifts by 10.3 ppm (C-3 of aglycone, from  $\delta$ 72.9 to 83.2) and 8.6 ppm (C-2 of glucose, from  $\delta$ 74.9 to 83.5) were observed. Therefore, in prosapogenin **2**, the O-3 of bayogenin should be glycosylated with  $\beta$ -D-glucopyranose bearing an  $\alpha$ -L-arabinopyranosyl unit on O-2.<sup>9,10</sup> The anomeric configurations of glucose and arabinose were confirmed as  $\beta$  and  $\alpha$ (L), respectively, based on the coupling constants of their anomeric protons;  $J=7.7\text{Hz}$ (glucose,  $\delta$ 5.09, d) and  $J=6.9\text{Hz}$ (arabinose,  $\delta$ 5.11, d). Hence, the structure of prosapogenin **2** was determined as 3-O-[ $\alpha$ -L-arabinopyranosyl(1 $\rightarrow$ 2)- $\beta$ -D-glucopyranosyl]-bayogenin.

A cellulase treatment of **1** gave prosapogenin **6** and xylose.<sup>11</sup> It was clearly shown that **6** still possesses three ester carbonyl groups( $\delta$ 176.0, 171.2 and 171.1) as does **1**. Treatment of **6** with 5%  $\text{K}_2\text{CO}_3$  afforded a deacylated compound **7** containing only one ester carbonyl group ( $\delta$ 176.1, C-28 of bayogenin). Compound **7** gave a  $^{13}\text{C}$ -NMR spectrum that was almost superimposable on that of **2** except that the C-28 of bayogenin was shifted and two extra sugar moieties, arabinose and rhamnose, were present. Furthermore, comparison of the  $^{13}\text{C}$ -NMR spectrum of **7** with that of [ $\alpha$ -L-rhamnopyranosyl(1 $\rightarrow$ 2)- $\alpha$ -L-arabinopyranosyl] 3-O-acetyl-oleanolate(**8**),<sup>12</sup> in which the carbons of the sugar moieties has been assigned, allowed us to deduce the sequence of the sugar chain on the C-28 carbonyl group of **7** to be an  $\alpha$ -L-rhamnopyranosyl-(1 $\rightarrow$ 2)- $\alpha$ -L-arabinopyranosyl group.

A mild alkaline hydrolysis of **1** with 0.5% KOH gave dibasic acid **10** and deacylated compound **9**, needles from MeOH, mp 233-5°C. The  $^{13}\text{C}$ -NMR spectrum of **9** was devoid of signals corresponding to the dicrotalic acid moiety. A glycosidation shift (**7** vs. **9**) by 10.8 ppm (from  $\delta$ 72.3 to 83.1) at C-3 of rhamnose indicating that the terminal xylose was attached through O-3 of rhamnose.<sup>10</sup> The anomeric configuration of the xylose was deduced to be  $\beta$  from the coupling constant of its anomeric proton ( $\delta$ 5.24, d,  $J=7.6\text{Hz}$ ) as well as the  $^{13}\text{C}$ -NMR. Thus the structure of **9** can be characterized as 3-O-[ $\alpha$ -L-arabinopyranosyl(1 $\rightarrow$ 2)- $\beta$ -D-glucopyranosyl]-bayogenin 28-[ $\beta$ -D-xylopyranosyl(1 $\rightarrow$ 3)- $\alpha$ -L-rhamnopyranosyl(1 $\rightarrow$ 2)- $\alpha$ -L-arabinopyranosyl] ester. The dibasic acid **10**, FAB-MS,  $m/z$  163(M+H) and 185(M+Na),  $^1\text{H}$ -NMR, two singlets at  $\delta$ 2.50 and 1.22, was crystallized from MeOH as white plates, mp 106-8°C, and identified as dicrotalic acid.<sup>6</sup>



- 1** R=  $\beta$ -D-Xyl  
**6** R= H



- 2** R= H  
**7** R=  $\alpha$ -L-Rha(1 $\rightarrow$ 2)- $\alpha$ -L-Ara-  
**9** R=  $\beta$ -D-Xyl(1 $\rightarrow$ 3)- $\alpha$ -L-Rha(1 $\rightarrow$ 2)- $\alpha$ -L-Ara-

Table 1. Carbon Chemical Shifts of Tubeimoside I, 1, and Related Compounds (in Pyridine-*d*<sub>5</sub>)

C.No.	Me.Bay <sup>a</sup>	1	9	6	7	2	8 <sup>12</sup>	Me.Gly <sup>b</sup>
Bayogenin Moiety								
1.	44.8	44.1	44.1	44.7	44.1	44.0		
2.	71.5	69.1	70.8 <sup>g</sup>	69.2	70.4 <sup>m</sup>	70.3		
3.	72.9	83.2	82.6 <sup>h</sup>	83.9	82.7 <sup>n</sup>	83.2 <sup>P</sup>		
4.	42.4	43.0	42.7	43.3	42.7	42.3		
5.	48.0	47.7	47.7	47.6	47.7	47.9		
6.	18.2	18.7	18.5	18.7	18.5	18.1		
7.	32.8	33.1	33.0	33.0	33.0	33.2		
8.	39.7	40.1	40.0	40.0	40.0	40.1		
9.	48.4	47.7	47.7	47.6	47.7	47.9		
10.	37.1	37.2	36.9	37.2	36.9	37.0		
11.	23.9	23.9	24.0	23.8	24.0	23.9		
12.	123.1	123.2	123.1	123.1	123.1	123.1		
13.	144.1	144.2	144.2	144.0	144.2	144.9		
14.	42.0	41.9	42.3	41.9	42.3	42.5		
15.	28.0	28.2	28.3	28.8	28.5	28.3		
16.	23.3	22.8	23.2	23.8	23.7	23.9		
17.	46.9	47.0	47.3	47.0	47.3	46.7		
18.	41.8	41.3	41.7	41.4	41.8	42.1		
19.	46.0	46.0 <sup>c</sup>	46.3	45.9 <sup>k</sup>	46.3	46.7		
20.	30.8	30.7	30.9	30.6	30.9	30.9		
21.	33.9	34.3	34.2	33.9	34.2	34.0		
22.	32.8	32.2	32.7	32.1	32.7	33.2		
23.	67.5	64.7 <sup>d</sup>	64.6	65.7	66.1	65.1		
24.	14.6	15.5	14.7	15.7	14.6	14.6		
25.	17.2	17.7	17.5 <sup>i</sup>	17.7 <sup>j</sup>	17.5 <sup>o</sup>	17.5 <sup>q</sup>		
26.	17.2	17.7	17.3 <sup>j</sup>	17.5 <sup>j</sup>	17.3 <sup>o</sup>	17.2 <sup>q</sup>		
27.	26.2	26.0 <sup>e</sup>	26.1	26.3	26.1	26.3		
28.	177.9	175.9	176.2	176.0	176.1	180.0		
29.	33.1	33.1	33.1	33.0	33.1	33.2		
30.	23.6	23.7	23.7	23.6	23.7	23.9		
Dicrotallic Acid Moiety								
1.		171.2		171.2				
2.		47.0 <sup>c</sup>		47.4 <sup>k</sup>				
3.		70.1		70.1				
4.		46.8 <sup>c</sup>		46.0 <sup>k</sup>				
5.		171.1		171.1				
3-Me		26.3 <sup>e</sup>		26.4				
Saccharide Chains								
Glc								
1.		103.5	103.5	103.6	103.3	103.4		105.4
2.		80.2	83.5 <sup>h</sup>	80.2	83.4 <sup>n</sup>	83.5 <sup>P</sup>		74.9
3.		78.8 <sup>f</sup>	78.0 <sup>j</sup>	79.0	77.8	77.9		78.1
4.		71.3	71.7	71.5	71.3	71.8		71.4
5.		78.3 <sup>f</sup>	78.2 <sup>j</sup>	78.3	78.0	78.3		78.1
6.		62.4	62.4	62.7	62.4	62.9		62.5
Ara								
1.		104.4	106.5	104.7	106.3	106.6		105.9
2.		73.7	73.8	73.8	73.8	73.8		72.2
3.		72.3	74.3	72.6	74.3	74.5		74.4
4.		72.3	69.2	72.6	69.2	69.2		69.1
5.		64.7 <sup>d</sup>	67.2	64.4	67.2	67.2		66.6
Ara								
1.		94.1	93.5	94.4	93.4		93.3	105.9
2.		74.6	75.6	74.9	75.2		74.9	72.2
3.		70.8	70.1 <sup>g</sup>	70.9	70.1 <sup>m</sup>		70.2	74.4
4.		67.6	66.2	67.5	66.1		66.0	69.1
5.		64.3 <sup>d</sup>	63.1	64.4	62.9		62.9	66.6
Rha								
1.		100.6	101.3	100.6	101.3		101.2	102.4
2.		72.3	71.2	72.6	72.3		72.1	72.2
3.		77.8 <sup>f</sup>	83.1 <sup>h</sup>	70.1	72.3		72.4	72.5
4.		73.2	72.8	75.4	73.7		73.7	73.6
5.		67.8	70.5 <sup>g</sup>	68.0	70.3 <sup>m</sup>		70.2	69.4
6.		18.2	18.5	18.4	18.5		18.4	18.4
Xyl								
1.		106.4	107.1					106.1
2.		74.6	75.0					74.6
3.		78.5 <sup>f</sup>	78.2 <sup>j</sup>					78.1
4.		71.1	71.0 <sup>g</sup>					70.9
5.		66.8	67.2					66.9

a: Bayogenin Methyl ester, b: Methyl Glycosides, (Skeletal carbons only)  
c - p: assignments might be interchangeable.

In order to determine the linkage positions of dicrotalic acid, a comparative  $^1\text{H-NMR}$  study was carried out on **6** and its deacylated derivative **7**. All proton signals of the sugar moieties of **6** and **7** were assigned using the COSY technique. Significant downfield shifts<sup>13</sup> of the H-4 signals of rhamnose and the terminal arabinose by 1.57ppm (from  $\delta$ 4.32 to 5.89) and 1.23 ppm (from  $\delta$ 4.33 to 5.56), respectively, indicated that the dicrotalic acid was linked to the O-4 of rhamnose and the O-4 of the terminal arabinose. This was corroborated by a set of acylation shifts<sup>11</sup> in  $^{13}\text{C-NMR}$ (**6** vs. **7**) involving C-3(upfield by 1.7 ppm from  $\delta$ 74.3 to 72.6), C-4 (downfield by 3.4 ppm from  $\delta$ 69.2 to 72.6) and C-5 (upfield by 2.8 ppm from  $\delta$ 67.2 to 64.4) of the terminal arabinose. A similar shift pattern was observed with the C-3, 4 and 5 of rhamnose. Therefore, it is obvious that one end of the dicrotalic acid bridge is bound to the O-4 of the terminal arabinose, and the other end, to the O-4 of rhamnose. Thus, the structure of **1** was determined unambiguously. Due to the influence of the dicrotalic acid bridging in the molecule, some difficulties were experienced to assign the carbon signals of **1** and its prosapogenin **6**, by comparing them with similar compounds with no inter-saccharidechain bridging. Therefore, C-H COSY experiments had to be performed on **6** to obtain the assignment of the carbon signals of the sugar moieties. Two notable upfield shifts by 3.2ppm(C-2 of glucose, from  $\delta$ 83.4 to 80.2) and 1.6 ppm(C-1 of arabinose, from  $\delta$ 106.3 to 104.7) were observed(**6** vs. **7**). These can be explained in terms of orientational changes around the glycosidic linkage(s) in the structure of **6**, and also in **1**, caused by the inter-saccharidechain bridge. It should be noted that in special cases like these, much care should be taken when employing empirical glycosidation and acylation shift rules for sequencing of sugar chains by  $^{13}\text{C-NMR}$ .

It is noteworthy that purified tubeimoside I showed moderate antitumor activity in primary *in vivo* pharmacological test.<sup>14</sup>

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